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An unusual synthesis of tetrahydrobenzo[*f*]isoquinolines^{\ddagger}

Ramendra Pratap,^a Resmi Raghunandan,^b P. R. Maulik^b and Vishnu Ji Ram^{a,*}

^aMedicinal and Process Chemistry Division, Central Drug Research Institute, Lucknow 226 001, India ^bMolecular and Structural Biology Division, Central Drug Research Institute, Lucknow 226 001, India

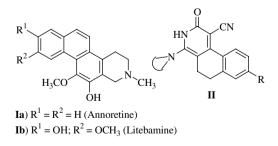
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Abstract—A facile and short synthesis of 2-oxo-4-*sec*-amino-2,3,5,6-tetrahydrobenzo[*f*]isoquinoline-1-carbonitriles has been delineated through base catalyzed ring transformation of 5,6-dihydro-2-oxo-4-*sec*-amino-2*H*-benzo[*h*]chromene-3-carbonitrile by cyanoacetamide in excellent yields.

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1. Introduction

Isoquinoline ring systems have been identified as major structural motifs, widely present in various natural products of chemical and therapeutic importance.¹ Annoretine² (**Ia**) and litabamine³ (**Ib**) are the two phenanthrene alkaloids isolated from Annota Montana and Litsea cubeba, respectively, in Figure 1, in which the former exhibits cytotoxic activity⁴ against different human cell cultures whilst the latter is active as a platelet antiaggregant⁵ and against acetyl choline esterase.⁶ These alkaloids belong to the family of [2,1-*f*]isoquinolines and, surprisingly, have received little attention towards the development of their chemistry. An extensive litera-





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2623405; e-mail: vjiram@yahoo.com

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ture survey revealed that the chemistry of highly functionalized benzo[*f*]isoquinoline ring systems is meagerly explored.

These diverse pharmacological activities and limited synthetic procedures inspired us to develop a novel and facile synthetic route, which could provide compounds with the benzo[*f*]isoquinoline motif.

A multistep synthesis of [2,1-f]isoquinolines has been reported⁷ from [2-(2-styrylphenyl)ethyl]methylcarbamic acid ethyl ester through a Bischler-Napieralski cyclization followed by photocyclization. Pyrolysis of 4-(2ethynylphenyl)pyridine at 810 °C and 0.5 mbar pressure resulted in benzo[f]isoquinolines as the major product.⁸ This procedure is only of theoretical importance. Cyclization of stilbenes by tributyltin hydride and AIBN in toluene at 80 °C produced⁹ benzo[f]isoquinoline as a major product. Junjappa and co-workers¹⁰ have also reported the synthesis of this class of compounds from the reaction of the oxoketene dithioacetal derived from 1-tetralone and enol acetals in the presence of phosphoric acid. A base catalyzed condensation of 1,2-(dibromomethylene)naphthalene with ethyl acetamidocyanoacetate is also an alternative route¹¹ for the construction of tetrahydrobenzo[*f*]isoquinolines. They have also been prepared¹² by electrocyclization of 1-vinyl-3,4-dihydronaphthalene-2-carboxaldehyde and N,N-dimethylhydrazones with the loss of dimethylamine. Recently, there is a thermolytic transformation of benzocyclobutenes into quinodimethane intermediates as dienes, which undergo intramolecular Diels-Alder cyclization reaction to produce¹³ benzo[*f*]isoquinolines. A new approach for

^{*} Corresponding author. Tel.: +91 522 2612411; fax: +91 522

the synthesis of 5,6-dihydrobenzimidazo[1,2-b]benzo[f]isoquinolines has been reported¹⁴ through condensation-cyclization of the oxoketene dithioacetal derived from 1-tetralone with 2-methyl/cyanomethylbenzimidazole but this procedure has the limitation of providing only methylsulfanyl substituted fused benzo[*f*]isoquinolines.

Herein, we report an efficient synthesis of 2-oxo-4-secamino-2,3,5,6-tetrahydrobenzo[f]-isoquinoline-1-carbonitriles 6 through base catalyzed ring transformation of 5.6-dihydro-2-oxo-4-sec-amino-2H-benzo[h]chromene-3-carbonitriles 4 by cyanoacetamide using powdered KOH as a base in DMF. However, we failed to effect the analogous ring transformation of 2-oxo-4-methylsulfanyl-5,6-dihydro-2H-benzo[h]chromene-3-carbonitriles 3 by cyanoacetamide, possibly due to the preference of a substitution reaction at C-4 of the chromene ring rather than the attack at C-10b for the ring transformation. To avoid these side reactions, 5,6-dihydro-2-oxo-4*sec*-amino-2*H*-benzo[*h*]chromene-3-carbonitriles were considered more appropriate precursors for the ring transformation reaction and were synthesized in two steps. The first step was the synthesis of 2-oxo-4-methylsulfanyl-5,6-dihydro-2H-benzo[h]chromenes 3 from the reaction of methyl 2-cyano-3,3-dimethylthioacrylate 1 and 1-tetralone 2 in DMSO using powdered KOH as a base. Amination¹⁵ of 3 was affected by refluxing with a secondary amine in ethanol (Scheme 1).

The topography of 5,6-dihydro-2-oxo-4-sec-amino-2Hbenzo[h]chromene-3-carbonitriles 4 revealed the presence of three electrophilic centre C-2, C-4 and C-10b in which the latter is highly prone to nucleophilic attack due to extended conjugation and the presence of an electron withdrawing CN substituent at C-3 position of the chromene ring. The nucleophile used as a carbanion was generated in situ from cyanoacetamide. Thus, a mixture of 4, cyanoacetamide 5 and powdered KOH in DMF was stirred at room temperature for 2–3 h. During this period all the precursors were consumed with the appearance of a new spot on TLC. The reaction mixture was poured onto crushed ice with vigorous stirring and thereafter neutralized with 10% HCl. The resulting

DMSO/KOH

5-6 h

2

Yield (%)

94

82

COOCH₃

3 R Н a

b

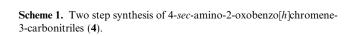
OCH₃

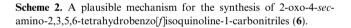
SCH₃

NC

H₃CS

1





precipitate was filtered, washed with water and dried. The crude products were purified by crystallization from chloroform-methanol and characterized as 2-oxo-4-secamino-2,3,5,6-tetrahydrobenzo[f]-isoquinoline-1-carbonitriles **6**.

A plausible reaction mechanism is shown in Scheme 2. The reaction is initiated by the attack of a carbanion generated in situ from cyanoacetamide at C-10b through the Michael addition followed by ring closure involving the amide nitrogen and C-4 of the chromene ring with the loss of carbon dioxide and acetonitrile to yield 6 (Scheme 2).

2. X-ray structural analysis

X-ray diffraction studies¹⁶ of 2-oxo-4-piperidin-1-yl-2,3,5,6-tetrahydrobenzo[f]isoquinoline-1-carbonitrile 6a revealed that the compound has a non-planarity induced chirality due to distortion in the aromatic rings. The conformation with arbitrary numbering is shown as an **ORTEP** diagram in Figure 2.

The presence of a cyano group in the bay region forces the molecule to buckle from the most favourable planar

CN NC DMF/KOH NH₂ ò ő ĊΝ JH н NH -CH₃CN NH ĊΝ 6

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R

Η

Η

H

Н

OCH₂

piperidin-1-vl

marpholin-4-yl

OCH₃ 4-methyl-piperidin-1-yl

piperidin-1-yl

4-methyl-piperidin-1-yl

4-benzyl-piperidin-1-yl

tetrahydroisoquinolin-2-yl

4-benzyl-piperazine-1-yl

6

я

b

с d Η

е Η

f

g

h

SCH3

C₂H₅OH

ΗN

4

CN

Yields (%)

95

92

97

95

91

90

97

89

Duration

2

2

2

2.5 1.5

1.5

2

1.5

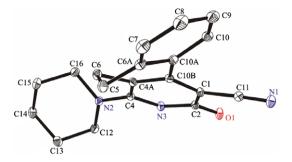


Figure 2. Displacement ellipsoid plot (30% probability) showing the molecular structure of 6.

arrangement through the bending of the aromatic rings out of the plane. The cyano group and the rings are also in different planes and in this situation the molecule acquires a helical conformation. The average mean plane angle for the twist between the terminal rings is 33.22° and the distance between the non-bonded carbon atoms in two molecules present in an asymmetric unit is 3.060 Å and 3.093 Å.

The X-ray studies further showed that the terminal rings are nearly planar while the central ring adopts a half chair conformation with a torsion angle C1–C10B– C10A–C10, -34.28° . It has structural analogy with the X-ray structure of methyl 1-(4-chlorophenyl)-3hydroxy-7-methoxy-9,10-dihydrophenanthrene-4-carboxylate.¹⁷ The torsional angle C4–C4A–C4B–C5 in the phenanthrene-4-carboxylate of the previously reported compound is 28.24° and the twist between the terminal rings is 28° while the distance between the non-bonded carbon atoms is 3.025 Å. From these observations, it is inferred that the presence of an isoquinoline ring with cyano substituent increases the degree of distortion and non-planarity induced helicity in the molecule (Fig. 3).

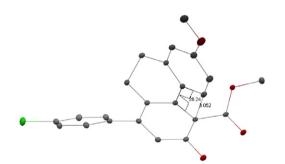


Figure 3. Methyl 1-(4-chlorophenyl)-3-hydroxy-7-methoxy-9,10-dihydrophenanthrene-4-carboxylate.

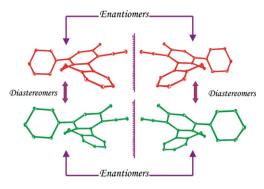


Figure 5. Structures of the benzo[*f*]isoquinoline showing the presence of both enantiomers and diastereomers.

The benzo[f]isoquinoline **6a** crystalized in a space group of C2/c. The crystal packing reveals the presence of two non-equivalent molecules in the asymmetric unit, Figure 4.

There are 16 molecules in the unit cell with two pairs of P and M enantiomers as shown in Figure 5. Out of the 16 molecules, 8 conformers have same bond distances as well as torsion angles, while the other 8 have different bond distances as well as torsion angles. This confirmed the presence of both enantiomeric as well as diastereomeric pairs in the unit cell.

Semi-empirical calculations were also performed for both benzo[f]isoquinoline and methyl 1-aryl-3-hydroxy-5,6-dihydrophenanthrene-4-carboxylate. Theoretical calculations were performed in order to study the influence of carbonyl as well as CN/COOCH₃ substituents. The semi-empirically optimized structures were then compared with the X-ray crystal structure. The structural parameters matched excellently with those previously obtained through X-ray structure analyses. All the molecules were fully optimized by the AM1 method. The geometry optimization showed that in the presence of a CN substituent in the bay region of benzo[*f*]isoquinoline, the torsion angle C1-C10B-C10A-C10 is 33.11° whereas the experimental value is -34.28° . When CN is replaced by COOCH₃ the torsion angle decreases to 28° both experimentally and theoretically. The theoretical calculation showed that the torsion angles of phenanthrene derivatives with CN and COOCH3 substituents in the bay region are 29° and 28°, respectively. The theoretical calculation and X-ray structure analysis revealed that in presence of a CN substituent in the bay region increases the distortion and nonplanarity of the rings.

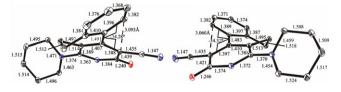


Figure 4. Diastereomers of the benzo[/]isoquinoline present in the asymmetric unit showing bond distances and torsion angles.

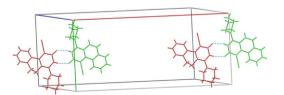


Figure 6. Crystal packing of benzo[f]isoquinoline showing the presence of intermolecular N-H···O interactions.

The crystal structure of benzo[*f*]isoquinoline demonstrated supramolecular assembly through intermolecular N–H···O interactions [N3–O1:2.785 Å; H3–O1:1.94 Å and N3–H3–O1:167°] with the formation of an eightmembered ring as indicated in Figure 6.

All the compounds synthesized were characterized by spectroscopic analysis.¹⁸

In summary, this is the first protocol for an efficient and convenient synthesis of 2-oxo-4-*sec*-amino-2,3,5,6-tetra-hydrobenzo[f]isoquinoline-1-carbonitriles **6** through base catalyzed ring transformation of 5,6-dihydro-2-oxo-4-*sec*-amino-2H-benzo[h]chromene-3-carbonitrile by cyanoacetamide.

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References and notes

- 1. Bentley, K. Nat. Prod. Rep. 2000, 17, 247.
- Wu, Y.-C.; Chang, G.-Y.; Duh, C.-Y.; Wang, S. K. Phytochemistry 1993, 33, 497.
- Wu, Y.-C.; Liou, J. Y.; Lee, S.-S.; Lu, S. J. Tetrahedron Lett. 1991, 32, 4169.
- (a) Whaley, W. M.; Meadow, M. J. Org. Chem. 1954, 19, 661; (b) Galiazzo, G.; Bortolus, P.; Masetti, F. J. Chem. Soc., Perkin Trans. 2 1975, 1712; (c) Masetti, F.; Bartocci, G.; Mazzucato, U. Gazz. Chim. Ital. 1982, 112, 255; (d) Tanga, M. J.; Almquist, R. G.; Smith, T. J. J. Heterocycl. Chem. 1985, 22, 1597.
- 5. Wu, Y.-C.; Liou, J. Y.; Duh, C.-Y.; Lee, S.-S.; Lu, S. T. J. *Pharmacol.* **1997**, *49*, 706.
- Chiou, C.-M.; Kang, J.-J.; Lee, S.-S. J. Nat. Prod. 1998, 61, 46.
- Martinez, E.; Estévej, J. C.; Estévej, R. J.; Castedo, L. *Tetrahedron* 2001, 57, 1973–1979.
- Dix, I.; Doll, C.; Hopf, H.; Jones, P. G. Eur. J. Org. Chem. 2002, 2547–2556.
- 9. Michael, J. P. Nat. Prod. Rep. 2002, 19, 742-760.
- Gupta, A. K.; Illa, H.; Junjappa, H. Tetrahedron 1990, 46, 2561–2572.
- (a) Wang, C.; Moseberg, H. I. *Tetrahedron Lett.* **1995**, *36*, 3623–3626;
 (b) Kotha, S.; Sreenivasachary, N. J. Indian Inst. Sci. **2001**, *81*, 277–286.

- Gilchrist, T. L.; Healy, M. A. M. Tetrahedron 1993, 49, 2543–2556.
- 13. Verrat, C.; Hoffmann, N.; Pete, J. P. Synlett 2000, 1166–1168.
- Panda, K.; Suresh, J. R.; Illa, H.; Junjappa, H. J. Org. Chem. 2003, 68, 3498–3506.
- 15. Pratap, R.; Ram, V. J. Tetrahedron Lett. 2007, 48, 2755.
- 16. The crystal data of **6a**: $C_{21.71}H_{21.7J}N_{3.43}O_{1.14}$, M = 349.00, monoclinic, C2/c, a = 28.997(3) Å, b = 13.1900(1) Å, c =18.057(2) Å, $\beta = 116.480(1)^{\circ}$, V = 6181.7(1) Å³, Z = 16, $D_c = 1.312 \text{ g cm}^{-3}$, μ (Mo-K α) = 0.083 mm⁻¹, F(000) = 2592, rectangular block, yellow, size = $0.5 \times 0.16 \times 10^{-10}$ 0.345 mm, 6350 reflections measured ($R_{int} = 0.0187$), 5445 unique, $wR_2 = 0.1448$ for all data, conventional $R = 0.0501 \left[(\Delta/\sigma)_{\text{max}} = 000 \right]$ on F-values of 3615 reflections with $I \ge 2\sigma(I)$, S = 0.944 for all data and 415 parameters. Unit cell determination and intensity data collection $(2\theta = 50^{\circ})$ were performed on a Bruker P4 diffractometer at 293(2) K. Structure solutions by direct methods and refinements by full-matrix least-squares methods on F^2 . Programs: XSCANS [Siemens Analytical X-ray Instrument Inc.: Madison, Wisconsin, USA 1996], SHELXTL-NT [Bruker AXS Inc.: Madison, Wisconsin, USA 1997]. CCDC (Deposit No: 661402) contains the supplementary crystallographic data. These data can be obtained free of charge from www.ccdc.cam.uk/conts/ retrieving.html [or from the Cambridge Crystallographic Data Center, 12 Union Road, Cambridge, CB2 1EZ, UK; fax: (internat.) +44 1223/336 033; e-mail: deposit@ccdc. cam.ac.uk.
- Sarkhel, S.; Goel, A.; Ram, V. J.; Chaudhary, S.; Maulik, P. R. Acta Crystallogr. 1997, C53, 1713.
- 18. General procedure for the synthesis of 2-oxo-4-sec-amino-2,3,5,6-tetrahydrobenzo[f]isoquinoline-1-carbonitriles 6: A mixture of 4 (0.5 mmol), cyanoacetamide 5 (0.6 mmol) and powdered KOH in DMF was stirred at room temperature for 2-3 h. During this period all the starting materials were consumed with the appearance of a new spot on TLC. Thereafter the reaction mixture was poured on crushed ice with vigorous stirring followed by neutralization with 10% HCl. The resulting precipitate was filtered, washed with water, dried and purified through neutral alumina column chromatography using chloroform as an eluent. Compound (6a): Yield 95%; mp >250°C; ¹H NMR (CDCl₃, 300 MHz) δ 1.74–1.76 (m, 6H, CH₂), 2.48–2.52 (m, 2H, CH₂), 2.71–2.75 (m, 2H, CH₂), 3.36–3.38 (m, 4H, CH₂), 7.26-7.29 (m, 1H, ArH), 7.37-7.39 (m, 2H, ArH), 8.17-8.22 (m, 1H, ArH); ¹³C NMR (CDCl₃, 75 MHz): δ 22.5, 22.9, 24.6, 28.4, 49.3, 86.6, 104.5, 117.1, 125.6, 126.3, 129.3, 130.1, 138.1, 152.6, 153.1, 162.8; IR (KBr): 2214 (CN) cm⁻¹; mass (ESI-MS) m/z 306 [M⁺⁺¹]; HRMS (70 eV): M⁺ Calcd for C₁₉H₁₉N₃O 305.15281, found: 305.15269.